

Supplementary Data

Enhanced Selectivity and Stability of Ruthenium Purple-Modified Carbon Fiber Microelectrodes for Detection of Hydrogen Peroxide in Brain Tissue

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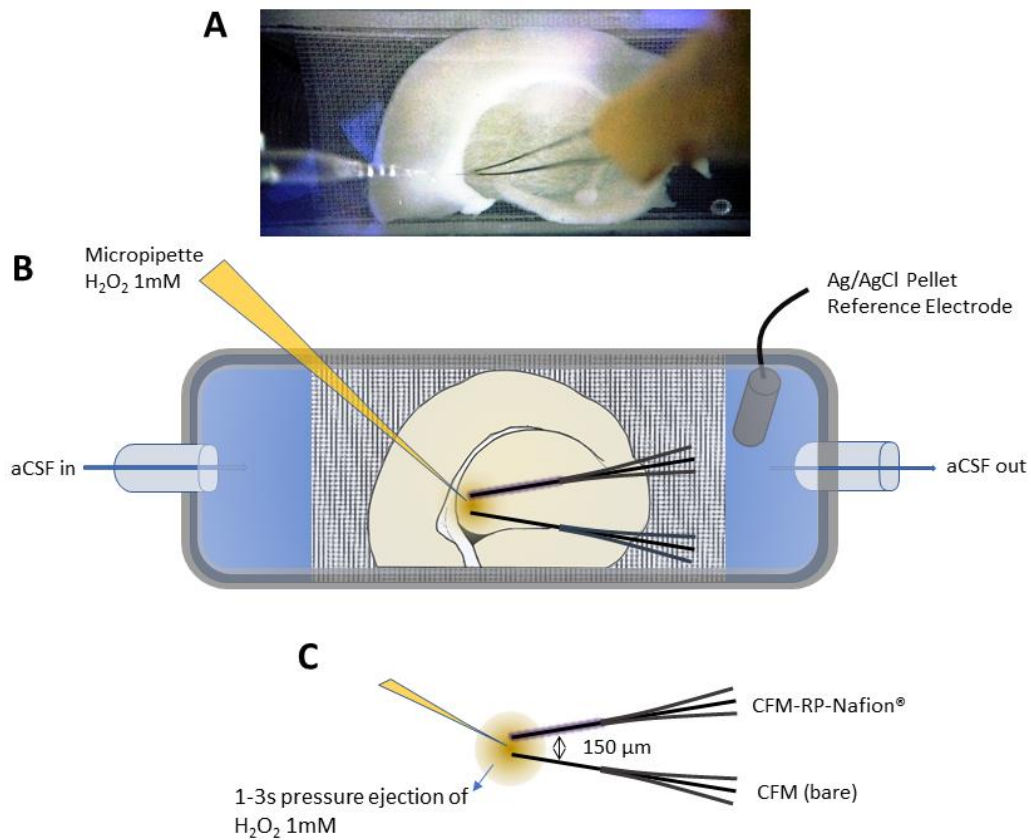


Figure S1 – Experimental setup for electrochemical recording in striatal slices. A) Photograph of the slice in the perfusion chamber showing an array of two microelectrodes (right) and a glass micropipette (left) inserted into the dorsal medial striatum; B) schematic representation of the striatal slice in the recording chamber, perfused with aCSF at 32 °C, showing microelectrode array (right) and micropipette (left); C) Detail of microelectrode and micropipette arrangement.

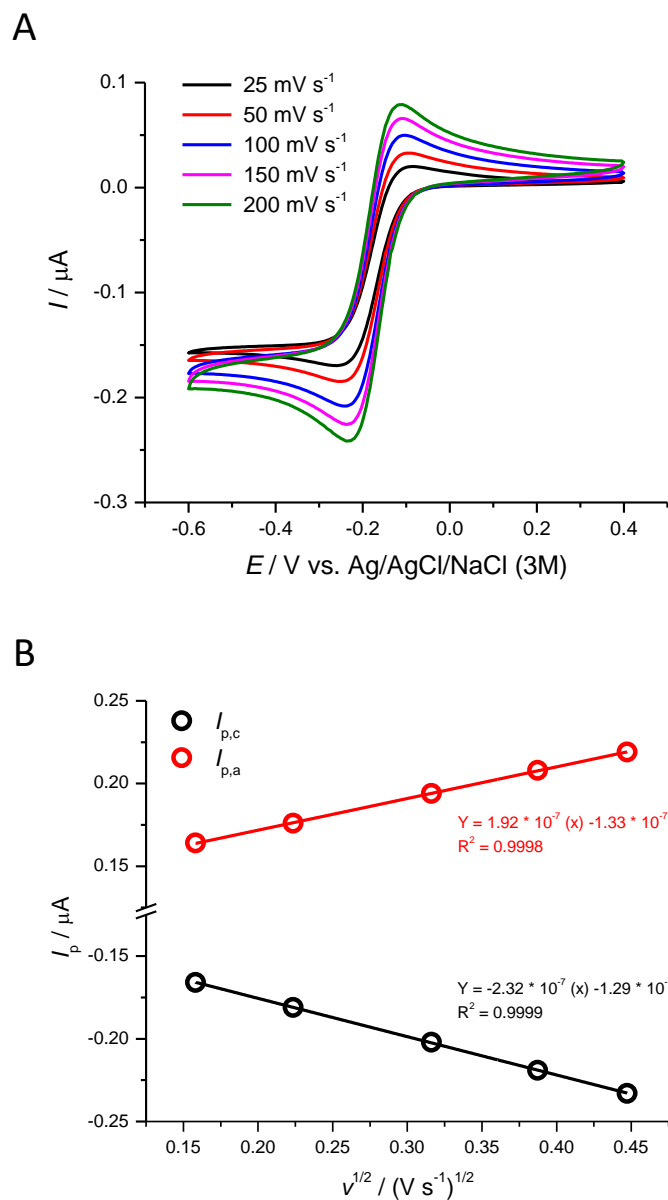


Figure S2 – Reversible Redox Reaction of Ru(III) (NH₃)₆ in 0.5 M KCl at increasing scan rates (A) and respective I_p vs. $v^{1/2}$ plot for determination of the electrochemical surface area of the CFM.

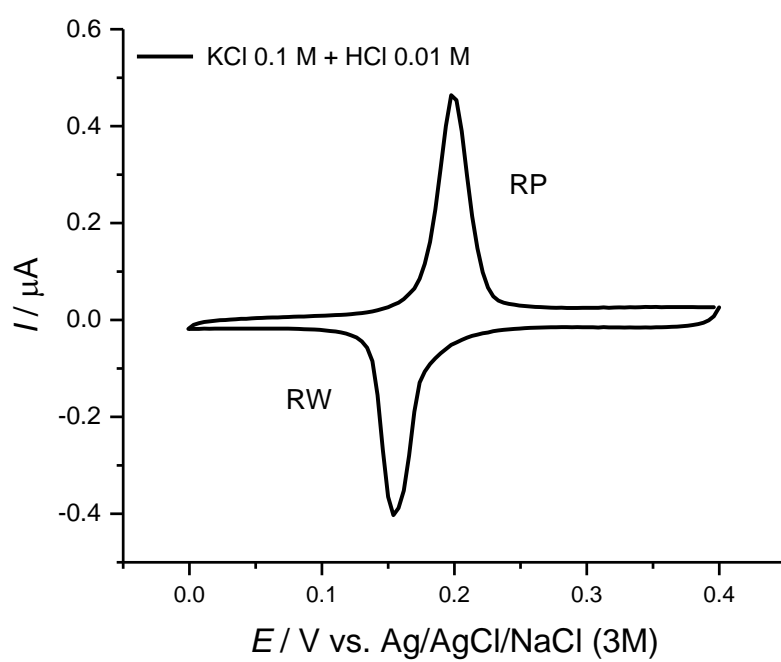
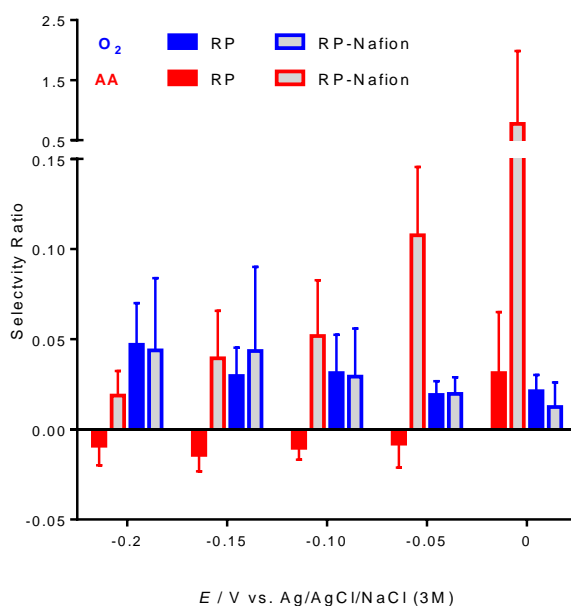


Figure S3 – Cyclic Voltammogram of a CFM-RP obtained in 0.1 M KCl + 0.01 M HCl, at a scan rate of 50 mV s^{-1} . RP – ruthenium purple; RW – ruthenium white.

A



B

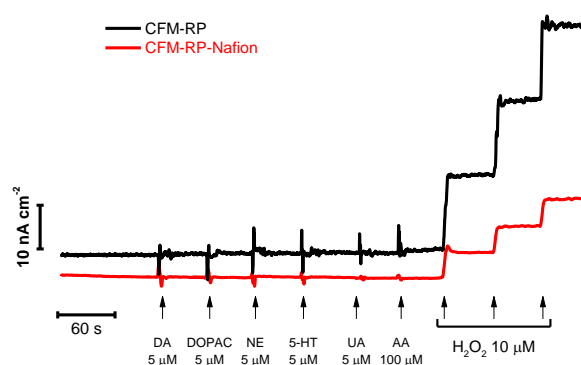


Figure S4 – Selectivity ratio for O₂ and ascorbate as a function of applied potential, calculated for CFM-RP and CFM-RP-Nafion® (A). Amperometric recording for a CFM-RP (black) and CFM-RP-Nafion® (red) at optimal applied potential (-0.1 and -0.2 V vs. Ag/AgCl, respectively) showing no response for the successive addition of dopamine (DA), 3,4-Dihydroxyphenylacetic acid (DOPAC), norepinephrine (NE), serotonin (4-HT) uric acid (UA) and ascorbic acid. Addition of H₂O₂ produced the expected response (B).