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Underline revision

1 **Rapid and sensitive methodology for determination of ethyl carbamate in fortified wines**
2 **using microextraction by packed sorbent (MEPS) and gas chromatography with mass**
3 **spectrometric detection (GC-MS)**

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12

13 **Abstract**

14 This work presents a new methodology to quantify ethyl carbamate (EC) in fortified wines. The
15 presented approach combines the microextraction by packed sorbent (MEPS), using a hand-held
16 automated analytical syringe, with one-dimensional gas chromatography coupled with mass
17 spectrometry detection (GC–MS). The performance of different MEPS sorbent materials was
18 tested, namely SIL, C2, C8, C18 and M1. Also, several extraction solvents and the matrix effect
19 were evaluated. Experimental data showed that C8 and dichloromethane were the best
20 sorbent/solvent pair to extract EC. Concerning solvent and sample volumes optimization used in
21 MEPS extraction an experimental design (DoE) was carried out. The best extraction yield was
22 achieved passing 300 μL of sample and 100 μL of dichloromethane. The method validation was
23 performed using a matrix-matched calibration using both sweet and dry fortified wines, to
24 minimize the matrix effect. The proposed methodology presented good linearity ($R^2=0.9999$)
25 and high sensitivity, with quite low limits of detection (LOD) and quantification (LOQ), 1.5 and
26 $4.5 \mu\text{g L}^{-1}$, respectively. The recoveries varied between 97 and 106%, while the method
27 precision (repeatability and reproducibility) was lower than 7%. The applicability of the
28 methodology was confirmed through the analysis of 16 fortified wines, with values ranging
29 between 7.3 and 206 $\mu\text{g L}^{-1}$. All chromatograms showed good peak resolution, confirming its
30 selectivity. The developed MEPS/GC-MS methodology arises as an important tool to quantify
31 EC in fortified wines, combining efficiency and effectiveness, with simpler, faster and
32 affordable analytical procedures that provide great sensitivity without using sophisticated and
33 expensive equipment.

34

35 **Abbreviations**

36 EC, ethyl carbamate; MEPS, microextraction by packed sorbent; GC-MS, gas chromatography-
37 mass spectrometry; FW, fortified wine; DoE, experimental design; BIN, barrel insert needle;
38 IS, internal standard; ME, matrix effect; LOD, limit of detection; LOQ, limit of quantification.

39 **Keywords:** Ethyl carbamate; Wines; Microextraction by packed sorbent; Gas chromatography-
40 mass spectrometric detection

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42 **1 Introduction**

43 Ethyl carbamate (EC), also known as urethane, is the ester of carbamic acid
44 ($\text{H}_2\text{NCOOC}_2\text{H}_5$). It is known as a toxic compound and was re-classified in 2007 by the
45 International Agency of Research on Cancer (IARC) as a probably carcinogenic to humans
46 (Group 2A) [1].

47 EC is formed in small amounts in fermented or heated food, namely in alcoholic
48 beverages, including fortified wines. The EC formation in these foodstuffs is usually associated
49 with storage time and temperature [2]. This compound results from the reaction between ethanol
50 and nitrogenous compounds, like urea, citruline, hydrocyanic acid and N-carbamyl compounds
51 [3-5]. One of the most common pathways proposed to explain the development of EC in acid
52 media consists in the reaction of urea with ethanol [6]. The kinetics of this reaction is greatly
53 enhanced by the temperature increase [7, 8]. Urea and citruline can be detected in wine and are
54 both derived from the arginine metabolism during the fermentative processes [6, 8, 9]. Another
55 precursor of EC referenced in the bibliography is hydrogen cyanide derived from cyanogenic
56 glycosides, produced by several plant species, including *Vitis vinifera* L. [10, 11]. The
57 formation of EC via cyanide is mostly originated through procedures that include thermal
58 treatments, like distillation or baking [6].

59 The toxicological concerns led Canada to establish by the first time, in 1985, legislation
60 regulating the EC limit values in alcoholic beverages, namely in fortified wines to $100 \mu\text{g L}^{-1}$.
61 Other legal limits were also imposed: $30 \mu\text{g L}^{-1}$ for table wines, $150 \mu\text{g L}^{-1}$ for distilled spirits,
62 $200 \mu\text{g L}^{-1}$ for sake and $400 \mu\text{g L}^{-1}$ for fruit brandies and liqueurs [6, 12]. In Europe, only Czech
63 Republic follows the Canadian legislation for fortified wine [12].

64 The concerns raised by the toxicological aspects of EC together with the low
65 concentration levels ($\mu\text{g L}^{-1}$) found in wines, as well as the occurrence of interferences on
66 detection, has motivated several researchers to develop new methods to determine it in wines.
67 Several extraction and chromatographic techniques have been used, including continuous

68 liquid-liquid extraction (LLE) with Soxhlet apparatus [13], derivatization with 9-xanthinol
69 followed by high performance liquid chromatography (HPLC) with fluorescence detection [14]
70 and even LLE after derivatization, followed by gas chromatography coupled with mass
71 spectrometry detection (GC-MS) [15]. On the other hand, the reference method set by the
72 International Organisation of Vine and Wine (OIV) [17] uses solid phase extraction (SPE)
73 preceding GC-MS quantification [16]. Other methods also make use of SPE, but use gas
74 chromatography with mass spectrometry (MDGC/MS) [18] and liquid chromatography
75 with tandem mass spectrometry (LC-MS/MS) for detection [19]. Most of the methodologies
76 found in literature to quantify EC use gas chromatography, using LLE [13, 20, 21] and SPE [16-
77 18, 22, 23] as extraction techniques. Nevertheless, several efforts have also been done to
78 develop new methodologies to determine EC without using long procedures and hard-working
79 analyses, combining precision to high sensitivity. In this regard, headspace solid phase
80 microextraction (HS-SPME) has been gaining great highlighting [24-26] and alternative
81 methodologies has been proposed using the most recent identification and quantification
82 technology, such as gas chromatography with tandem mass spectrometry detection (GC-
83 MS/MS) [26] and two-dimensional gas chromatography with time-of-flight mass spectrometry
84 (GC×GC-ToFMS) [25]. Liao et al. [27] also used an emergent extraction technique, based on
85 ultrasound-assisted emulsification-microextraction (USAEME) to extract EC in alcoholic
86 beverages, but using gas chromatography coupled to triple quadrupole mass spectrometry.
87 However, this kind of technologies is still not accessible to many laboratories.

88 Recently, microextraction by packed sorbent (MEPS) has also becoming emergent,
89 arising as a feasible and easy-to-use extraction technique. MEPS derives from the
90 miniaturization of the conventional SPE, but with additional advantages: uses small sample and
91 solvent volumes (microliters) and consequently reduces the environmental impact, increases the
92 analysis sensitivity and enables the direct injection into the LC or GC instruments. The small
93 cartridge can be packed or coated with different silica-based polymers: SIL (unmodified silica),
94 C2 (ethyl), C8 (octyl), C18 (octadecyl) and M1 (80% C8 and 20% SCX - strong cation
95 exchanger using sulfonic acid bonded silica), providing selective and suitable sampling

96 conditions [28]. The MEPS technique has been used to determine other compounds of interest
97 for the alcoholic beverages industry [29-31], however, as far as we know, it has never been
98 applied for the analytical determination of EC.

99 The aim of this study was the development of a fast, simple and sensitive methodology
100 to quantify EC in fortified wines using MEPS extraction combined with one-dimensional GC-
101 MS equipment, accessible to most laboratories.

102 **2 Materials and methods**

103 **2.1 Chemicals and samples**

104 Ethyl carbamate (EC) was purchased from Acros Organics (Geel, Belgium), while butyl
105 carbamate (BC), used as internal standard (IS), was obtained from Sigma–Aldrich (Steinheim,
106 Germany). All standards had a purity grade of more than 97%. Absolute ethanol, > 99.8% (GC),
107 was purchased from Sigma–Aldrich (Steinheim, Germany), tartaric acid and methanol from
108 Panreac (Barcelona, Spain) while acetonitrile, ethyl acetate and dichloromethane were from
109 Fisher Scientific (Leicestershire, UK). Ultra-pure water (18 MΩ) was prepared by the
110 Simplicity®UV ultrapure water (type 1) apparatus from Millipore (Milford, MA, USA).

111 EC and BC stock solutions of 1 g L⁻¹ were prepared by dissolving appropriate amounts
112 of each compound in ultra-pure water. In order to obtain the matrix-matched calibration
113 solutions, suitable dilutions of the stock solutions were prepared with ultra-pure water, to obtain
114 the intermediate solutions of 50 mg L⁻¹ in EC and 10 mg L⁻¹ in BC, which were then used to
115 spike dry and sweet fortified young wines. Each calibration point was extracted in triplicate,
116 within the validation range 5-400 µg L⁻¹.

117 The sweet and dry fortified wines used to perform the matrix-matched calibrations were
118 obtained from *Vitis vinifera* L. white varieties and were absent of quantifiable amounts of EC
119 and BC. Regarding the application sample set, 16 fortified wines, aged up to 36 years old and
120 with ethanol contents between 18 to 20% were analyzed using the developed methodology.

121 **2.2 Apparatus and chromatographic conditions**

122 eVol® MEPS™ hand-held automated analytical syringe (SGE Analytical Science,
123 Australia) of 500 µL was used and MEPS barrel insert needles (BINs, 8 µL, 45 µm particle size
124 and 60 Å pore size), containing 4 mg of different packing polymers (SIL, C2, C8, C18 and M1)
125 were tested to optimize the extraction.

126 All analyses were carried out using a GC-MS system, the TRACE GC Ultra gas
127 chromatograph equipped with the ISQ single quadrupole and the TriPlus autosampler (liquid
128 mode) from Thermo Scientific (Hudson, NH, USA). The column was a DB-WAX 60 m × 0.250
129 mm with 0.50 µm film thickness from Agilent J&W (Folsom, CA, USA). The carrier gas was
130 helium at a constant flow rate of 1 mL min⁻¹. The injector port that was kept at 230 °C, in
131 splitless mode, while the transfer line and the ion source were maintained at 230 and 240 °C,
132 respectively. The oven temperature program started at 40 °C, hold 1 min, increased to 180 °C at
133 20 °C min⁻¹ and hold for 15 min, with a total GC run time of 23 min.

134 The mass spectrometer was operated in electron impact (EI) mode at 70 eV. Initially,
135 some tests with standards and samples were performed with chromatograms obtained in total
136 ion count (TIC), in the range m/z 30–400, to ensure the retention time of EC and BC. Then,
137 selective ion monitoring (SIM) of the three characteristic ions m/z 62, 74 and 89 of both
138 compounds was tested in order to ensure good resolution. Also, to increase the sensitivity and to
139 meet quantification purposes, further analyses were performed using the ion m/z 62.

140 **2.3 MEPS optimization**

141 As aforementioned, retention times of EC and BC were previously determined using
142 individual standards dissolved in dichloromethane, with chromatograms recorded in TIC.
143 Several solvents were individually analyzed in order to check the absence of EC, specifically
144 ethanol, methanol, acetone, ethyl acetate, acetonitrile and dichloromethane, through direct
145 injection into GC-MS. Additionally, several commercially available sorbent materials (SIL, C2,
146 C8, C18 and M1) were tested and the extraction was performed with all EC free solvents.
147 Meantime, the best extraction solvent was also chosen. The standard solution used for these
148 tests was set to 100 µg L⁻¹ of EC (limit imposed by Canada) spiked with 24 µg L⁻¹ of BC

149 (internal standard). After choosing the ideal BIN and extraction solvent, the MEPS procedure
150 was then optimized performing an experimental design (DoE). This is an experimental strategy
151 in which factors (experimental variables that can affect the response) are varied together, instead
152 of one at a time. The experiments carried out are designed economically and efficiently, while
153 individual and combined factors are evaluated [32]. In this study, the analyzed factors were the
154 sample and solvent volumes to be used in the extraction procedure. As response variable, the
155 GC-MS data was used, namely to evaluate the factors-levels combination that ensure its
156 maximization. For each factor, three levels were examined, varying from 200 – 1000 μL and
157 100 – 350 μL to sample and solvent volumes, respectively. The plan to carry out the
158 experiments as well as the data analysis was computed using Matlab software (version 7.6, the
159 Mathworks Inc.).

160 **2.4 MEPS optimized procedure**

161 Firstly, 5 mL of sample/standard solution, previously spiked with 12 μL of internal
162 standard (BC solution of 10 mg L^{-1}), were filtered through 0.45 μm syringe Acrodisc GHP
163 filters (Pall Gelman Sciences, Ann Arbor, MI, USA). Following this step, samples were then
164 extracted using the C8 sorbent, which was selected to extract EC, after being performed the
165 optimization tests. Before each extraction, the sorbent was washed and conditioned twice with
166 500 μL of methanol, dichloromethane and ultra-pure water, at about 33 $\mu\text{L s}^{-1}$. Then, 300 μL of
167 sample were passed through the sorbent at a flow rate of about 5 $\mu\text{L s}^{-1}$. Thereafter, a drying
168 step was performed passing, five-fold, 500 μL of air at 250 $\mu\text{L s}^{-1}$. EC was then eluted with 100
169 μL of dichloromethane, aspirating at 1.7 $\mu\text{L s}^{-1}$ and dispensing at 33 $\mu\text{L s}^{-1}$, approximately. Each
170 sample/standard solution was extracted in triplicate and 3 μL of extract were injected twice into
171 the GC-MS port. Each BIN was used for about 120 extractions. The DoE optimized MEPS
172 extraction procedure is schematized in Fig. 1.

173 **2.5 Method validation**

174 The described MEPS/GC-MS methodology for determination of EC in fortified wines
175 was validated in terms of linearity, sensitivity, matrix effect, selectivity, precision and accuracy.

176 The working standard solutions were prepared by spiking both dry and sweet fortified
177 wines at six different concentration levels: 5, 10, 50, 100, 200 and 400 $\mu\text{g L}^{-1}$ of EC with 24 μg
178 L^{-1} of BC as internal standard. Calibration curves were obtained by plotting the analyte peak
179 area ratio (EC area/IS area) from the six increasing standard solutions against the corresponding
180 EC concentration. The linearity (R^2) was determined based on the linear regression results.

181 Sensitivity was evaluated determining the limit of detection (LOD) and limit of
182 quantification (LOQ) as follow: $LOD=3.3 \sigma/b$ and $LOQ=10 \sigma/b$, with σ as the intercept standard
183 deviation and b the slope.

184 The matrix effect (ME) was assessed through the percentage of the quotient between the
185 slopes of the curves obtained from the standards solutions in synthetic wine (6 g L^{-1} of tartaric
186 acid, 18% of ethanol and pH 3.50) and those obtained by spiking dry and sweet fortified wines
187 with known amounts (matrix-matched calibration), by the following equation [33]:

$$188 \quad \% \text{ ME} = \left[\frac{(\text{slope of matrix-matched calibration} - \text{slope of synthetic wine calibration})}{\text{slope of synthetic wine calibration}} \right] \times 100$$

189 Selectivity was appraised by the analysis of several fortified wines, among which were
190 chosen those that were used for the matrix-matched calibration, to ensure the absence of
191 chromatographic interferences, at the retention times of EC and BC (SIM at m/z 62), which
192 could compromise EC quantification. Synthetic wine blanks were also evaluated.

193 Precision was estimated from inter- and intra-day analysis of the standard solutions and
194 fortified wines. Intra-day repeatability was assessed by 10 successive replicate determinations of
195 2 samples and a working standard solution, while inter-day reproducibility was assessed by the
196 analyses of the same samples in 3 different days. These two parameters were expressed as
197 relative standard deviation (%RSD).

198 The accuracy of the method was assessed through a recovery study, spiking a fortified
199 wine in triplicate, with known amounts of EC at three representative concentrations levels,

200 within the calibration range. Average recovery was calculated by comparing mean values of the
201 3 replicates with theoretical concentrations of each one. Carry-over was also investigated by
202 running a blank sample after extracting the working standard solutions with the highest content
203 of EC.

204 **3 Results and discussion**

205 Firstly, a concentrated solution of EC, diluted in dichloromethane, was directly injected
206 into the GC-MS and recorded at full scan mode (total ion count) to identify and determine its
207 retention time (t_R). Then, several ramp temperatures were tested in order to optimize the GC-MS
208 analysis of EC. At the same time, to ensure the absence of interfering substances at EC retention
209 time (14.1 min), some non-optimized MEPS extracts of fortified wine samples were analyzed
210 with both TIC and SIM modes. At SIM mode, the analyses were performed recording the sum
211 of the three major ions m/z 62, 74 and 89 and also, only the characteristic ion m/z 62. It was
212 found interferences at the EC retention time when the recording was done with the sum of the
213 ions m/z 62, 74 and 89. Indeed, the TIC mode analysis confirmed that the matrix of some
214 fortified wines was very complex and concentrated, compromising the sensitivity. In this sense,
215 it was chosen to perform SIM analysis only at m/z 62, which assured enough sensitivity to
216 analyze EC with an excellent performance. Similar strategy has already been adopted by other
217 authors [18, 34, 35].

218 **3.1 Extraction solvent survey**

219 Taking into account the objective of developing an extraction method with MEPS, the
220 potential extraction solvents were analyzed looking for the presence of EC, with the SIM mode
221 at m/z 62. The obtained results showed that only acetonitrile, ethyl acetate and dichloromethane
222 were EC free solvents. Methanol, ethanol and acetone solvents had measurable amounts of EC,
223 mainly ethanol, which presented the peak with the greatest area of EC.

224 This result led us to avoid the use of this solvent for calibration purposes, considering
225 that EC presence in ethanol could affect its quantification. Actually, the EC determination in

226 alcoholic beverages presupposes its use to simulate the matrix, since ethanol has direct influence
227 on the EC extraction. Thus, this fact must be taken into account on the development of
228 analytical methods, once standards solutions are currently prepared with a certain percentage of
229 ethanol [25, 26, 36-38].

230 Actually, we have tried to find an ethanol source that was absent of EC. In this sense, 3
231 bottles of ethanol > 99.8% (GC) of the same brand, available in the laboratory, were tracked
232 through GC-MS direct injection, in order to quantify EC. In this case, the calibration was
233 prepared based on standards diluted in dichloromethane. The resulting calibration showed good
234 linearity and sensitivity results ($R^2 = 0.9999$ and $LOQ = 15.21 \mu\text{g L}^{-1}$). The results revealed that
235 ethanol bottles presented concentrations ranging from 25.0 to 27.9 $\mu\text{g L}^{-1}$.

236 Therefore, the synthetic wine, usually used for the preparation of standards, can have an
237 additional EC concentration of about 4.5 $\mu\text{g L}^{-1}$ derived from the added ethanol (18%).

238 **3.2 Selection of the MEPS sorbent and extraction solvent**

239 After solvents survey, several tests were conducted in order to select the best
240 solvent/sorbent pair. To perform this task, a non-optimized MEPS procedure was carried out
241 using a 500 μL syringe coupled with a hand-held automatic system. This syringe was fitted with
242 a removable BIN containing 4 mg of sorbent material. The performance of the sorbent materials
243 SIL, C2, C8, C18 and M1 were tested with the 3 extraction solvents free of EC. Very recently,
244 new sorbents became commercially available, which could be further tested.

245 The obtained results were compared to select the best BIN and solvent to extract and
246 quantify EC in fortified wines. Regarding the extraction solvent, it was verified that acetonitrile
247 extracts less EC compared to the other two extraction solvents, regardless the BIN used.
248 Moreover, acetonitrile extracted some interfering substances that co-eluted with the EC and BC
249 peaks (Fig. 2). In turn, ethyl acetate and dichloromethane were the solvents with higher
250 efficiency in the extraction of EC and BC. Actually, ethyl acetate extracts more EC than
251 dichloromethane, however, causes a change in the baseline, reducing the signal to noise ratio
252 (S/N) of both EC and BC peaks (Fig. 2). Furthermore, the EC peak of the ethyl acetate extracts

253 presented an inferior resolution, as depicted in Fig. 2. Considering these results,
254 dichloromethane was chosen as extraction solvent.

255 The BIN with C8 sorbent material presented the best efficiency to extract EC (Fig. 2),
256 using dichloromethane as extraction solvent. Thus, C8 BIN and dichloromethane were chosen
257 to perform the MEPS/GC-MS methodology for the determination of EC in fortified wines.

258 **3.3 MEPS extraction optimization**

259 After choosing the C8/dichloromethane pair, an experimental design (DoE) was carried
260 out to optimize the extraction in order to obtain the best response in the GC-MS equipment. The
261 sample and extraction solvent volumes were the chosen variables. The sample volumes
262 analyzed were 200, 500 and 1000 μL , while the tested solvent volumes were 100, 200 and 350
263 μL . Fig. 3 depicts the result of the statistical DoE approach. The two factors analyzed were
264 plotted against the response variable in order to visualize the combination that maximizes the
265 GC-MS response. Moreover, the response of other interferences was also analyzed in order to
266 ensure that the chosen factors combination maximize the S/N of the methodology used.

267 The optimum conditions were achieved by maximizing the second order function,
268 which has sample and solvent volume as dependent variables and GC response as independent
269 variable. As illustrated by Fig. 3, the maximum EC peak area can be achieved by using 100 μL
270 of dichloromethane and 300 μL of wine sample. Other conditions that also affect the MEPS
271 extraction, such as aspiration/dispense rates and conditioning/equilibration steps, were adjusted
272 (section 2.4) taking into account the tips reported by previous methods, ensuring efficiency and
273 effectiveness [39, 40].

274 **3.4 Matrix effects**

275 The matrix effect can compromise the results generated by an analytical method,
276 especially when it is intended to analyze samples of high complexity, such as fortified wines.
277 Thus, the variation percentages of the slopes of three calibration curves, accessed with synthetic,

278 dry and sweet fortified wines as samples matrix and using the optimized extraction, were
279 compared to evaluate the matrix influence on the extraction procedure and analysis.

280 Although there is no limit values established for matrix effect, it can be considered that
281 up to 15% of matrix suppression or enhancement is acceptable. In the present study a value of
282 17% was obtained, revealing a small matrix effect when wines are used instead of synthetic
283 wine. A negligible difference was found (about 0.3%) between the two types of wines.

284 **3.5 Method validation**

285 Faced with the lack of an ethanol completely free of EC, together with the fact that was
286 observed matrix effect, it was decided to adopt the matrix-matched calibration approach to
287 overcome these drawbacks. To accomplish this calibration, the selectivity of the proposed
288 methodology was firstly assessed by the analysis of the sweet and dry fortified wines, which
289 were further used to generate the matrix-matched calibration. The results revealed that there
290 were no significant interferences at EC and BC retention times, 14.1 and 19.4 min, respectively,
291 as demonstrated in Fig. 4.

292 A single calibration curve was then obtained by the average response of the six
293 concentration levels prepared with both sweet and dry fortified wine standard solutions. Each
294 one was extracted in triplicate and injected in duplicate. A good correlation coefficient
295 ($R^2=0.9999$) was observed, confirming the linearity of the method. Table 1 depicts some of the
296 validation results.

297 The method sensitivity was evaluated by LOD and LOQ determinations, calculated
298 based on the obtained linear regression (section 2.5). The LOD and LOQ were low (1.5 and 4.5
299 $\mu\text{g L}^{-1}$, respectively), being close or even lower to those found in literature [3, 13, 15, 18, 19, 24-
300 26, 34, 35, 37, 41], conferring to the developed methodology a great sensitivity to analyze EC in
301 fortified wines.

302 Recovery study was carried out to determine the accuracy of the method, by spiking a
303 fortified wine with known amounts of EC, at three concentration levels representative of the
304 calibration range. The wine sample was analyzed before and after the addition of 3 different

305 amounts of EC. The recoveries ranged between 97 and 106%, demonstrating the good accuracy
306 of the developed methodology (Table 1).

307 The method precision (repeatability and reproducibility) was evaluated by the variation
308 of intra- and inter-day (three different days with an interval of 5 days between them) repetition
309 method. Repeatability was accessed by 5 successive extractions injected twice of $100 \mu\text{g L}^{-1}$
310 standard solution and 2 fortified wines, with different concentrations. The reproducibility was
311 estimated by the variation between the intra-day results and those obtained in inter-day analyses,
312 through the extraction (triplicate) and injection (duplicate) of the same 3 samples. The results
313 revealed a good repeatability (5 – 7%) and reproducibility (4 – 7%) of the methodology, since
314 all RSD values were lower than 7%, regardless the area and height of the EC peak.

315 Additionally, the analysis of blanks after extracting the standard solutions with the
316 highest content of EC, confirmed the absence of carry-over between extractions.

317 **3.6 Analysis of fortified wine samples**

318 To evaluate the applicability of the proposed MEPS/GC-MS methodology for
319 determination of EC in fortified wines a set of fortified wines, aged up to 36 years old, were
320 analyzed. All samples were extracted in triplicate and injected twice. The results are shown in
321 Table 2. The older wines were analyzed in order to check the adopted linear range, as EC
322 content is expected to increase with age [2].

323 The obtained chromatograms showed that the applicability of the MEPS/GC-MS
324 methodology to quantify EC in fortified wines was achieved, since they showed a good peak
325 resolution, confirming its selectivity. Additionally, the quantified concentrations varied from 7.3
326 to $206 \mu\text{g L}^{-1}$, showing that the developed methodology covers the range interest of the
327 compound (Table 2). Actually, the fact that wines with higher content of EC were in general
328 associated with higher ageing periods was also demonstrated.

329 **4. Conclusion**

330 A fast, simple and sensitive methodology was developed and optimized to quantify EC
331 in fortified wines using MEPS extraction, through a hand-held automated analytical syringe,
332 with GC–MS detection. The best solvent/sorbent pair was selected after testing several sorbent
333 materials and EC free extraction solvents. C8 BIN and dichloromethane were the most efficient
334 pair to extract EC. MEPS extraction was optimized performing an experimental design, varying
335 sample and extraction solvent volumes. The best response could be achieved with the passage of
336 300 μL of sample and 100 μL of dichloromethane.

337 The matrix effect study revealed that a noticeable effect of both sweet and dry fortified
338 wines exists relative to synthetic wine. In turn, and together with the fact that it was not found
339 an ethanol completely free of EC commercially available, a matrix-matched calibration was
340 performed using both sweet and dry fortified wines. The analytical methodology was then
341 validated, showing good results in terms of linearity, sensitivity, selectivity precision and
342 accuracy. The applicability of the methodology was demonstrated by the analysis of a set of 16
343 fortified wines, with values ranging between 7.3 and 206 $\mu\text{g L}^{-1}$. The corresponding
344 chromatograms showed good precision and resolution.

345 Finally, it can be concluded that the presented MEPS/GC-MS methodology is an
346 excellent tool to quantify EC in fortified wines, gathering efficiency and effectiveness, without
347 using long and hard-working procedures, like the conventional methodology adopted by the
348 OIV.

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352 **References**

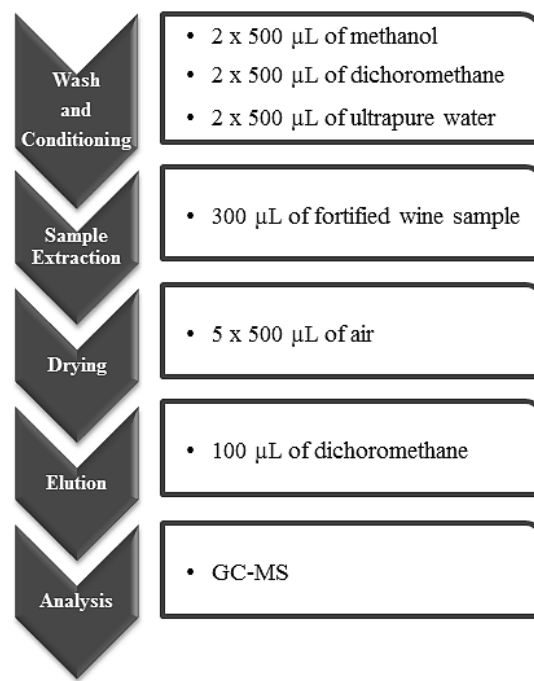
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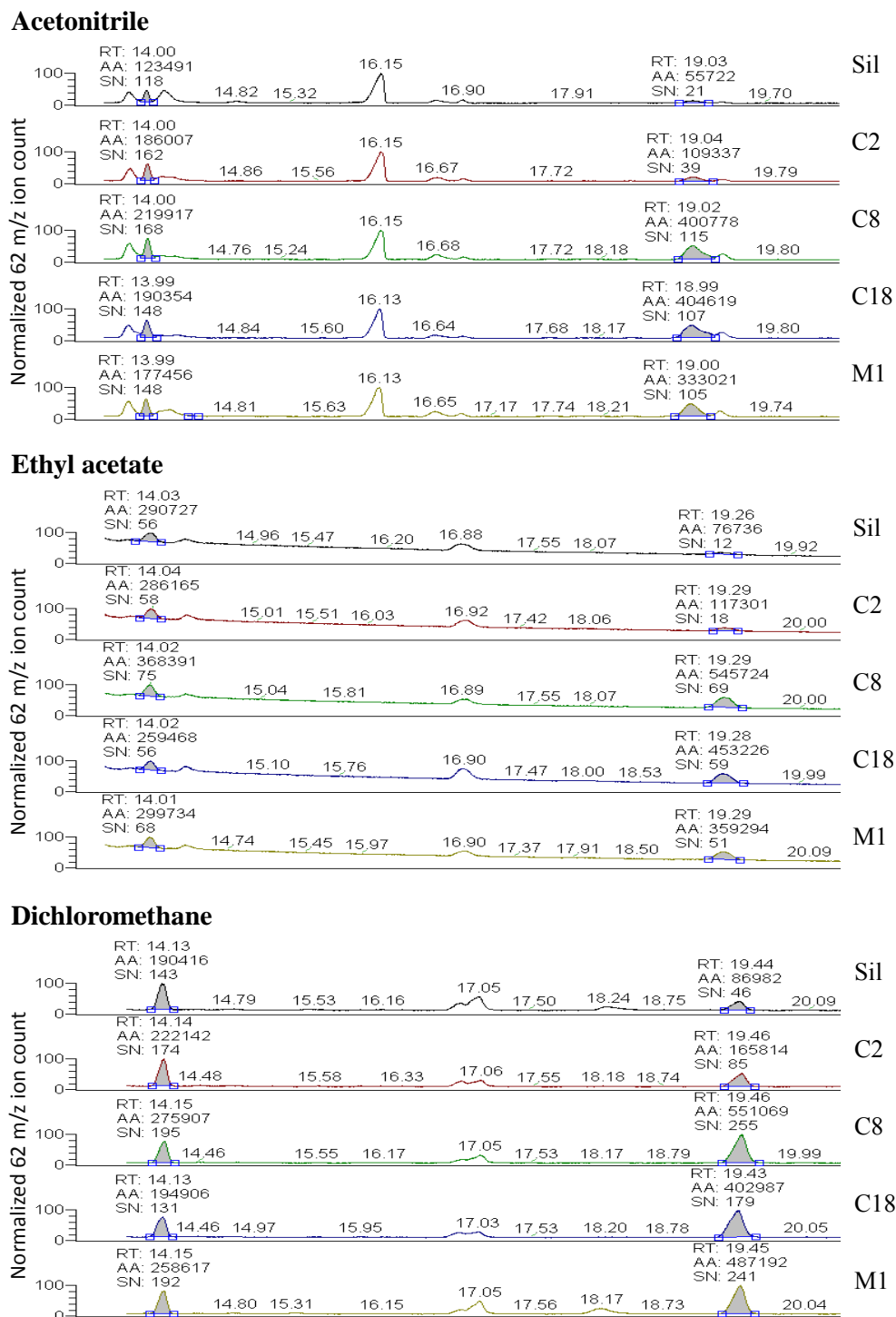
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428 **Figures**

429

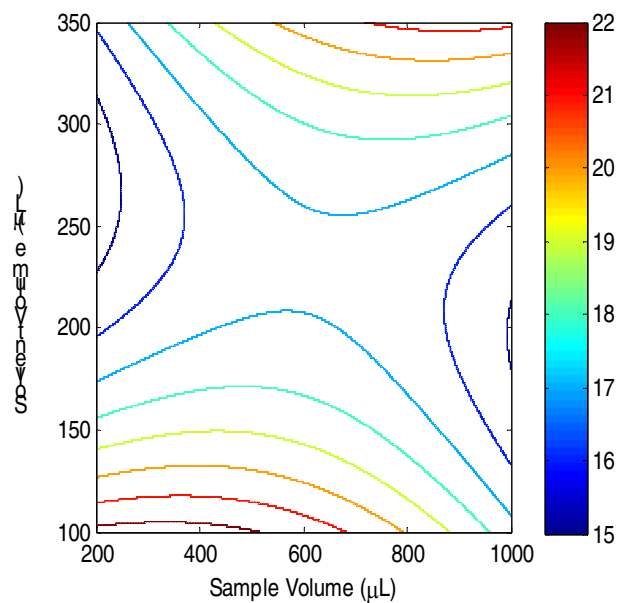
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Fig. 1. DoE optimized MEPS procedure for determination of EC in fortified wines.



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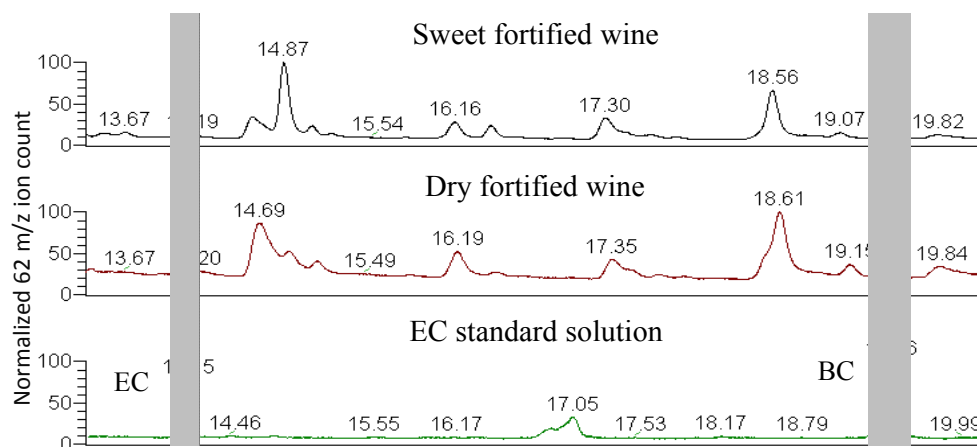
432 **Fig. 2.** Typical chromatograms of the sorbent materials SIL, C2, C8, C18 and M1 using the
 433 extraction solvents acetonitrile, ethyl acetate and dichloromethane. EC retention time \approx 14 min,
 434 BC retention time \approx 19 min. RT - retention time; AA – peak area; SN - signal to noise ratio.



435

436 **Fig. 3.** DoE to optimize the MEPS extraction with C8 BIN with sample volume, extraction
437 solvent volume and the response in GC-MS equipment as variables. The colormap illustrates the
438 variation of GC-MS response, where the maximum is delimited by the dark red line.

439



440

441 **Fig. 4.** Chromatograms of the fortified wine samples used to generate the matrix-matched
 442 calibration and a $100 \mu\text{g L}^{-1}$ standard solution of EC with $24 \mu\text{g L}^{-1}$ of BC. EC – ethyl
 443 carbamate; BC – butyl carbamate.

444 **Tables**

445 **Table 1.** Validation results obtained for the proposed MEPS/GC-MS methodology.

Parameter	Result	
linear regression ($y=mx+b$)	$0.01045x + 0.13741$	
Linear concentration range	$5\text{-}400 \mu\text{g L}^{-1}$	
R^2	0.9999	
LOD ($\mu\text{g L}^{-1}$)	1.5	
LOQ ($\mu\text{g L}^{-1}$)	4.5	
Recovery	$C_c \pm SD (\mu\text{g L}^{-1})$	%
FW	26 ± 2	-
FW + EC $50 \mu\text{g L}^{-1}$	78 ± 4	106
FW + EC $100 \mu\text{g L}^{-1}$	123 ± 6	97
FW + EC $200 \mu\text{g L}^{-1}$	228 ± 10	101

LOD - limit of detection; LOQ- limit of quantification; C_c - Concentration; FW - fortified wine; SD - standard deviation

446

447 **Table 2.** Application of the proposed methodology for the EC quantification of 16 fortified

448 wines.

	Wine age (years)	Concentration ($\mu\text{g L}^{-1}$)	SD (n=6)
FW ₁	5	28	3
FW ₂	5	31	3
FW ₃	5	22	4
FW ₄	3	18	2
FW ₅	5	38	2
FW ₆	3	50	2
FW ₇	5	13	2
FW ₈	unk	7.6	0.1
FW ₉	17	76.1	0.7
FW ₁₀	16	85.5	0.9
FW ₁₁	36	132	5
FW ₁₂	18	138	5
FW ₁₃	18	107	3
FW ₁₄	17	93	3
FW ₁₅	25	206	7
FW ₁₆	unk	7.3	0.3

FW - fortified wine; unk - unknown; SD - standard deviation

449

450

450 **Highlights**

451 MEPS was firstly used to quantify ethyl carbamate in fortified wines.

452 The extraction was optimized (DoE) to 300 μL of sample and 100 μL of dichloromethane.

453 Good linearity ($R^2=0.9999$) and low LOQ ($4.5 \mu\text{g L}^{-1}$) were obtained.

454 The method applicability was demonstrated by the analysis of 16 fortified wines.

455 MEPS ensured efficiency and effectiveness without using sophisticated equipment.

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